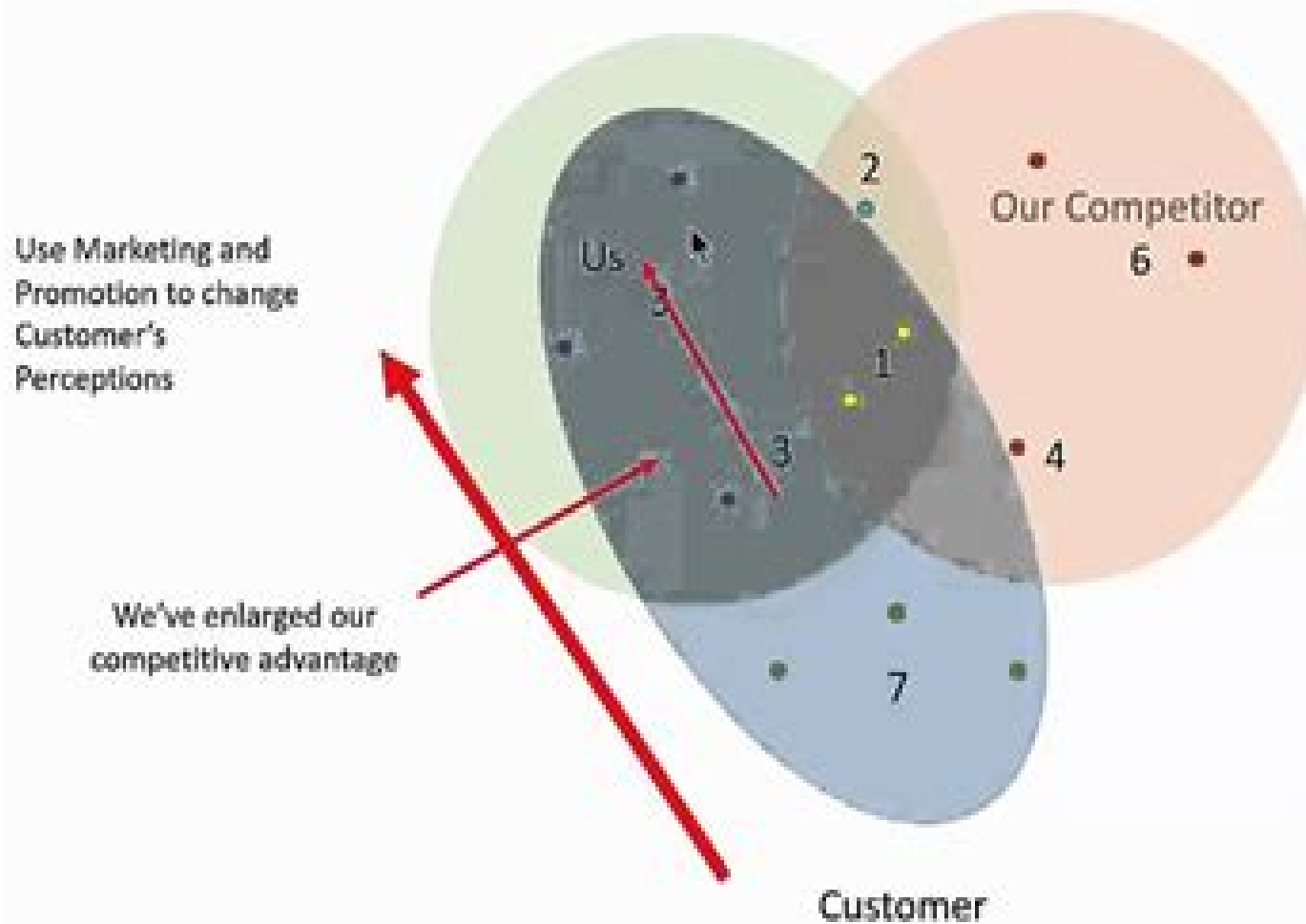


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Change Perceptions



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seaborgium determine if the following electron configurations are correct 26 1s²2s²2p⁶3s²3p⁶4s²3d¹⁰4p⁶5s¹ no it should be 3d¹⁰

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web how many protons neutrons and electrons are in atoms of these isotopes write the complete electron configuration for each isotope answer co has 27 protons 27 electrons and 33 neutrons 1s² 2s² 2p⁶ 3s² 3p⁶ 4s² 3d⁷ i has 53 protons 53 electrons and 78 neutrons 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 4d¹⁰ 5s² 5p⁵

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web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations of larger atoms question how are the electron configurations of elements beyond argon determined