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Systems for Maximum
Efficiency

ANDY TREMAYNE

Senior Principal, Oracle Applications Performance Group

FOREWORD BY CLIFF GODWIN

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web qualitative test flame tests introduction for this test i had five containers labelled a e each containing a metal salt i had to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and testing it by putting the metal under a roaring bunsen flame safety points wear googols all the time when

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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the

unknown slat burned orange but also had a slight blue color close to the stick

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions lino3 cu no3 2 sr no3 2 ba no3 2 kno3 and nano3 since nitrates do not emit color you

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web task use a flame test to determine which ion salt produces the red color then using a spectroscope match the bright line spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material list causes red fireworks procedure

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web energy and matter answered worksheet the experiment conducted aimed to give students the opportunity to determine different metal ions in liquid and solid compounds by observing the color of the flames they give off when

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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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