

Nutshell review & preview of ORGANIC REAGENTS

1. Alcoholic KOH

$R-X \rightarrow$ Alkene ; Elimination

2. Aluminium Ethoxide



(Aldehyde)

(Ester)

3. Aqueous KOH/NaOH

$R-X \rightarrow ROH$

Nucleophilic substitution reaction also used for Cannizzaro reaction with aldehyde.

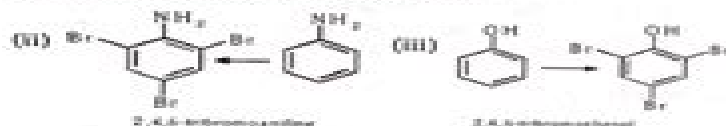
4. Baeyer's Reagent (Alkaline cold dilute $KMnO_4$)



alkene \longrightarrow 1, 2 diol
(used to detect unsaturation)

5. Bromine water

(i) used to detect unsaturation;



6. Benedict's solution

Used to detect aldehyde group $RCHO \rightarrow RCO_2^-$
[ketone gives -ve test]

7. $Cu_2Cl_2 + NH_4OH$

Used to Detect Terminal Alkyne
Red Precipitate observed

8. CrO_2Cl_2

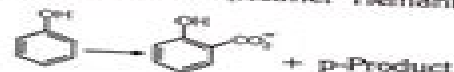


Etard reaction

9. CrO_3

- (i) $RCH_2OH \rightarrow RCHO$
- (ii) $R_2CHOH \rightarrow R_2C=O$
- (iii) $R_3COH \rightarrow$ no reaction

10. $CCl_4 + OH^-$ (Reimer Tiemann)



11. $CO + HCl + AlCl_3$



Gatterman koch reaction

12. $HCN + HCl + AlCl_3$



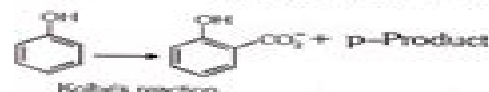
Gatterman Aldehyde Synthesis

13. $CHCl_3 + KOH$



(ii) $RNH_2 \rightarrow RNC$ (Carbyl amine reaction)
(used to detect 1° Amine) (Isocyanide test)

14. $CO_2 + OH^-$ (high temp. + Pressure)



Kolbe's reaction

15. Cu/Δ

- (i) $RCH_2OH \rightarrow RCHO$
- (ii) $R_2CHOH \rightarrow R_2C=O$



16. 2,4 - D.N.P

used to detect carbonyl group (orange ppt observed)

17. DMSO

Polar aprotic solvent: favour S_N2 mechanism

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