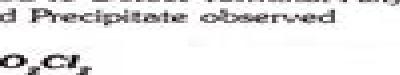


Nutshell review & preview of
ORGANIC REAGENTS

1. **Alcoholic KOH**
 $R-X \rightarrow \text{Alkene}$: Elimination
2. **Aluminium Ethoxide**
 $\text{RCH-O} \longrightarrow \text{R}-\underset{\substack{\text{O} \\ \parallel}}{\text{C}}-\text{O}-\text{CH}_2\text{R}$ (Tischenko Reaction)
 (Aldehyde) (Ester)
3. **Aqueous KOH/NaOH**
 $R-X \rightarrow \text{ROH}$
 Nucleophilic substitution reaction also used for Cannizzaro reaction with aldehyde.
4. **Baeyer's Reagent** (Alkaline cold dilute KMnO_4)
 $\text{RCH}=\text{CHR}' \longrightarrow \text{RCH}-\underset{\substack{\text{OH} \\ |}}{\text{C}}-\underset{\substack{\text{OH} \\ |}}{\text{CH}}-\text{R}'$
 (Syn)
 alkene \longrightarrow 1, 2 diol
 (used to detect unsaturation)
5. **Bromine water**
 (i) used to detect unsaturation:
 (ii) 
 2,4,4'-Biphenylamine
 (iii) 
 2,4,4'-Biphenol
6. **Benedict's solution**
 Used to detect aldehyde group $\text{RCHO} \rightarrow \text{RCO}_2^-$
 (ketone gives -ve test)
7. **$\text{Cu}_2\text{Cl}_2 + \text{NH}_4\text{OH}$**
 Used to Detect Terminal Alkyne
 Red Precipitate observed
8. **CrO_3Cl_2**

 Etard reaction
9. **CrO_3**
 (i) $\text{RCH}_2\text{OH} \rightarrow \text{RCHO}$
 (ii) $\text{R}_2\text{CHOH} \rightarrow \text{R}_2\text{C=O}$
 (iii) $\text{R}_3\text{COH} \rightarrow$ no reaction

10. $\text{CCl}_4 + \text{OH}^-$ (Reimer Tiemann)

11. $\text{CO} + \text{HCl} + \text{AlCl}_3$

Gattermann-Koch reaction

12. $\text{HCN} + \text{HCl} + \text{AlCl}_3$

Gattermann Aldehyde Synthesis

13. $\text{CHCl}_3 + \text{KOH}$

Reimer Tiemann reaction

(i) $\text{RNH}_2 \rightarrow \text{RNC}$ (Carbyl amine reaction)
(used to detect 1° Amine) (Isocyanide test)

14. $\text{CO}_2 + \text{OH}^-$ (high temp. + Pressure)

Kolbe-Kohlschmidt reaction

15. Cu/A

(i) $\text{RCH}_2\text{OH} \rightarrow \text{RCHO}$
(ii) $\text{R}_2\text{CHOH} \rightarrow \text{R}_2\text{C=O}$

(iii) $\text{H}_3\text{C}-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{C}}}-\text{OH} \longrightarrow \text{H}_3\text{C}-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{C}}}=\text{C}-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{CH}_3}}$

16. 2,4 - D.N.P.

used to detect carbonyl group (orange ppt observed)

17. **DMSO**

Polar aprotic solvent: favour $\text{S}_{\text{N}}2$ mechanism

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