

Organic Chemistry II Study Guide for Exam I
Chapters 10, 11, and 14

- Classification of Alcohols
 - Alcohols are classified according to how many carbons are attached to the carbon containing the -OH group.

- Methyl, primary, secondary, and tertiary

- Nomenclature of Alcohols

- Naming an alcohol

- Start numbering the carbon chain to reach the -OH group first
 - Include the location (carbon number) of the -OH group



- 2-methylhexan-2-ol

- Naming an alcohol as a substituent

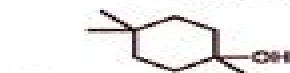
- Do this when alcohol is NOT the highest priority group (carboxylic acid, etc.)
 - Start numbering the carbon chain at the higher priority group
 - Use the prefix hydroxy- and include the location (carbon number) of the -OH group



- 4-hydroxybutanoic acid

- Naming cyclic alcohols

- If the -OH is the highest priority group, do NOT include the location (carbon number)
 - It is implied so it is unnecessary



- 1,4,4-trimethylcyclohexanol

- Naming diols

- When 2 -OH groups are present and is the highest priority group, use the suffix -diol
 - Include the location (carbon number) of BOTH -OH groups



- Propane-1,3-diol

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William John Keep



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1. Understanding the eBook Organic Chemistry Lecture Guide Part One
 - The Rise of Digital Reading Organic Chemistry Lecture Guide Part One
 - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry Lecture Guide Part One
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chemistry Lecture Guide Part One
 - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chemistry Lecture Guide Part One
 - Personalized Recommendations
 - Organic Chemistry Lecture Guide Part One User Reviews and Ratings
 - Organic Chemistry Lecture Guide Part One and Bestseller Lists
5. Accessing Organic Chemistry Lecture Guide Part One Free and Paid eBooks
 - Organic Chemistry Lecture Guide Part One Public Domain eBooks
 - Organic Chemistry Lecture Guide Part One eBook Subscription Services
 - Organic Chemistry Lecture Guide Part One Budget-Friendly Options
6. Navigating Organic Chemistry Lecture Guide Part One eBook Formats

- ePub, PDF, MOBI, and More
- Organic Chemistry Lecture Guide Part One Compatibility with Devices
- Organic Chemistry Lecture Guide Part One Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Organic Chemistry Lecture Guide Part One
 - Highlighting and Note-Taking Organic Chemistry Lecture Guide Part One
 - Interactive Elements Organic Chemistry Lecture Guide Part One
- 8. Staying Engaged with Organic Chemistry Lecture Guide Part One
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chemistry Lecture Guide Part One
- 9. Balancing eBooks and Physical Books Organic Chemistry Lecture Guide Part One
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chemistry Lecture Guide Part One
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chemistry Lecture Guide Part One
 - Setting Reading Goals Organic Chemistry Lecture Guide Part One
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chemistry Lecture Guide Part One
 - Fact-Checking eBook Content of Organic Chemistry Lecture Guide Part One
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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