

Organic Chemistry II Study Guide for Exam I  
Chapters 10, 11, and 14

- Classification of Alcohols
  - Alcohols are classified according to how many carbons are attached to the carbon containing the -OH group.

- Methyl, primary, secondary, and tertiary

- Nomenclature of Alcohols

- Naming an alcohol

- Start numbering the carbon chain to reach the -OH group first
    - Include the location (carbon number) of the -OH group



- 2-methylhexan-2-ol

- Naming an alcohol as a substituent

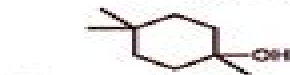
- Do this when alcohol is NOT the highest priority group (carboxylic acid, etc.)
    - Start numbering the carbon chain at the higher priority group
    - Use the prefix hydroxy- and include the location (carbon number) of the -OH group



- 4-hydroxybutanoic acid

- Naming cyclic alcohols

- If the -OH is the highest priority group, do NOT include the location (carbon number)
    - It is implied so it is unnecessary



- 1,4,4-trimethylcyclohexanol

- Naming diols

- When 2 -OH groups are present and is the highest priority group, use the suffix -diol
    - Include the location (carbon number) of BOTH -OH groups



- Propane-1,3-diol

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## **Table of Contents Organic Chem Study Pkg 4e**

1. Understanding the eBook Organic Chem Study Pkg 4e
  - The Rise of Digital Reading Organic Chem Study Pkg 4e
  - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chem Study Pkg 4e
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chem Study Pkg 4e
  - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chem Study Pkg 4e
  - Personalized Recommendations

- Organic Chem Study Pkg 4e User Reviews and Ratings
- Organic Chem Study Pkg 4e and Bestseller Lists
- 5. Accessing Organic Chem Study Pkg 4e Free and Paid eBooks
  - Organic Chem Study Pkg 4e Public Domain eBooks
  - Organic Chem Study Pkg 4e eBook Subscription Services
  - Organic Chem Study Pkg 4e Budget-Friendly Options
- 6. Navigating Organic Chem Study Pkg 4e eBook Formats
  - ePub, PDF, MOBI, and More
  - Organic Chem Study Pkg 4e Compatibility with Devices
  - Organic Chem Study Pkg 4e Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Organic Chem Study Pkg 4e
  - Highlighting and Note-Taking Organic Chem Study Pkg 4e
  - Interactive Elements Organic Chem Study Pkg 4e
- 8. Staying Engaged with Organic Chem Study Pkg 4e
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chem Study Pkg 4e
- 9. Balancing eBooks and Physical Books Organic Chem Study Pkg 4e
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chem Study Pkg 4e
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chem Study Pkg 4e
  - Setting Reading Goals Organic Chem Study Pkg 4e
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chem Study Pkg 4e
  - Fact-Checking eBook Content of Organic Chem Study Pkg 4e

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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