

Molecule **C** is propionaldehyde. The carbon connected to oxygen is  $sp^2$  hybridized. Please reconsider your answer.

**Page reference:** pages 73-76; Section 2.2

**Type:** multiple choice question

**Title:** Chapter 02 – Question 3

**03)** Which of the following compounds is a carboxylic acid?



**a. A**

**Feedback:** Incorrect. For a molecule to be classed as a carboxylic acid, the  $sp^2$ - hybridized carbon of the carbonyl must be bonded to OH and a carbon-based substituent ( $CH_3$ ).

Molecule **A** is ketone (acetone, aka propanone) because the  $sp^2$  hybridized carbon atom of the carbonyl forms a single bond to two carbon-based substituents. Please reconsider your answer.

**Page reference:** pages 77-78; Section 2.3

**\*b. B**

**Feedback:** Correct. Molecule **B** is a carboxylic acid (acetic acid, aka ethanoic acid) because the  $sp^2$ - hybridized carbon of the carbonyl is bonded to OH and carbon-based substituents ( $CH_3$ ).

**Page reference:** pages 77-78; Section 2.3

**c. C**

**Feedback:** Incorrect. For a molecule to be classed as a carboxylic acid, the  $sp^2$  hybridized carbon of the carbonyl must be bonded to OH and carbon-based substituents.

Molecule **C** is an acyl halide (acetyl chloride) because the  $sp^2$  hybridized carbon atom of the carbonyl forms a single bond to a halogen (Cl) and a carbon-based substituent ( $CH_3$ ). Please reconsider your answer.

**Page reference:** pages 77-78; Section 2.3

**d. D**

**Feedback:** Incorrect. For a molecule to be classed as a carboxylic acid, the  $sp^2$ -hybridized carbon of the carbonyl must be bonded to OH and a carbon-based substituent.

Molecule **D** is an aldehyde (acetaldehyde) because the  $sp^2$  hybridized carbon atom of the carbonyl forms a single bond to a halogen and a hydrogen. Please reconsider your answer.

**Page reference:** pages 77-78; Section 2.3

**Type:** multiple choice question

**Title:** Chapter 02 - Question 04

**04)** What is the IUPAC name for:



**a. 2-methyl-5-isobutylheptane**

**Feedback:** The IUPAC name of a compound is made up of three parts: the suffix, the parent chain and the prefix. The longest continuous carbon chain is called the parent chain. All alkane names end with -ane and this ending is the suffix used for all saturated hydrocarbons. Any substituents or minor functional groups are listed in the prefix.

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**Ralph J. Fessenden, Joan S. Fessenden**



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