

Molecule C is propionaldehyde. The carbon connected to oxygen is sp^2 hybridized. Please reconsider your answer.

Page reference: pages 73-76; Section 2.2

Type: multiple choice question

Title: Chapter 02 – Question 3

03) Which of the following compounds is a carboxylic acid?



a. A

Feedback: Incorrect. For a molecule to be classed as a carboxylic acid, the sp^2 - hybridized carbon of the carbonyl must be bonded to OH and a carbon-based substituent (CH_3).

Molecule A is ketone (acetone, aka propanone) because the sp^2 hybridized carbon atom of the carbonyl forms a single bond to two carbon-based substituents. Please reconsider your answer.

Page reference: pages 77-78; Section 2.3

b. B

Feedback: Correct. Molecule B is a carboxylic acid (acetic acid, aka ethanoic acid) because the sp^2 - hybridized carbon of the carbonyl is bonded to OH and carbon-based substituents (CH_3).

Page reference: pages 77-78; Section 2.3

c. C

Feedback: Incorrect. For a molecule to be classed as a carboxylic acid, the sp^2 -hybridized carbon of the carbonyl must be bonded to OH and carbon-based substituents.

Molecule C is an acyl halide (acetyl chloride) because the sp^2 hybridized carbon atom of the carbonyl forms a single bond to a halogen (Cl) and a carbon-based substituent (CH_3). Please reconsider your answer.

Page reference: pages 77-78; Section 2.3

d. D

Feedback: Incorrect. For a molecule to be classed as a carboxylic acid, the sp^2 -hybridized carbon of the carbonyl must be bonded to OH and a carbon-based substituent.

Molecule D is an aldehyde (acetraldehyde) because the sp^2 hybridized carbon atom of the carbonyl forms a single bond to a halogen and a hydrogen. Please reconsider your answer.

Page reference: pages 77-78; Section 2.3

Type: multiple choice question

Title: Chapter 02 - Question 04

04) What is the IUPAC name for:



a. 2-methyl-5-isobutylheptane

Feedback: The IUPAC name of a compound is made up of three parts: the suffix, the parent chain and the prefix. The longest continuous carbon chain is called the parent chain. All alkane names end with -ane and this ending is the suffix used for all saturated hydrocarbons. Any substituents or minor functional groups are listed in the prefix.

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Ralph J. Fessenden,Joan S. Fessenden

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