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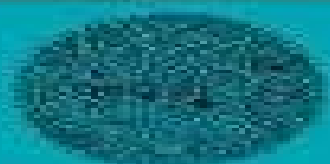
IBP Handbook No 8

Methods for Chemical Analysis of Fresh Waters

H. L. Golterman

Second Printing

$\text{Ca}^{2+} = 2.08$



$\text{Mg}^{2+} = 0.70$ meq/l

$\text{NH}_4^+ = 0.05$ meq/l

$\text{Fe}^{2+} = 0.10$ meq/l



$\text{K}^+ + \text{Na}^+ = 0.45$ meq/l

Total = 4.22 meq/l

$\text{HCO}_3^- = 3.54$ meq/l

$\text{SO}_4^{2-} = 0.32$ meq/l

$\text{Cl}^- = 0.25$ meq/l

$\text{HSiO}_3^- = 0.10$ meq/l

$\text{PO}_4^{3-} \text{ as } \text{P} = 0.01$ meq/l

Total = 4.25 meq/l



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