

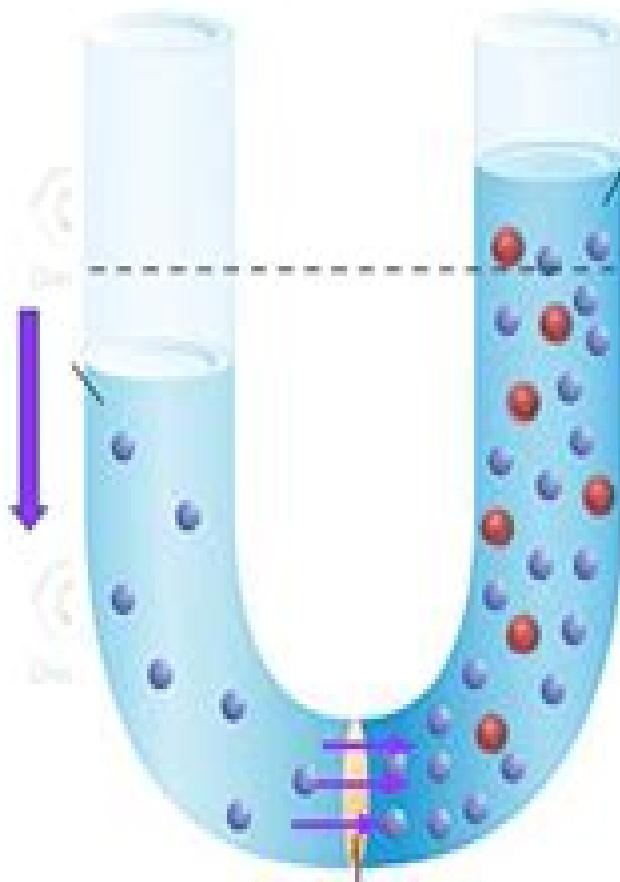
The Osmotic Pressure

Solvent

Solution

Solvent

Solution



Solvent molecules are passing through the semipermeable membrane raising the level of the solution.

Solvent molecules are trying pass through the semipermeable membrane by osmosis.

Solvent



External pressure to stop osmosis.

*The pressure, required to stop the osmosis, is called the **osmotic pressure**:*

Osmotic Pressure

Paul Brown Dunbar

Osmotic Pressure:

Osmotic Pressure Alexander Findlay,1913 **The Osmotic Pressure of Cane-sugar Solutions in the Vicinity of the Freezing Point of Water** ... William West Holland,1907 Osmotic Pressure Measurements of Glucose Solutions at 100, and 200 ... Walter Hatheral Coolidge,1915 *The Osmotic Pressure of Aqueous Solutions* Harmon Northrop Morse,1914 The Osmotic Pressure of Mannite Solutions Between Ten and Forty Degrees ... Edwin Louis Frederick,1915

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