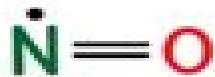


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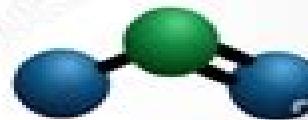
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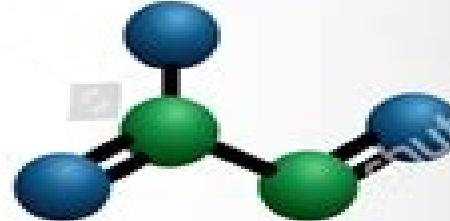
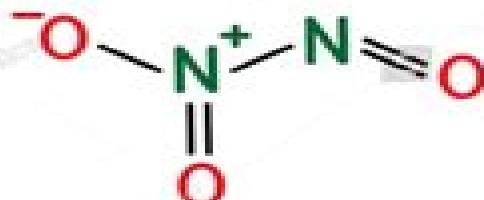
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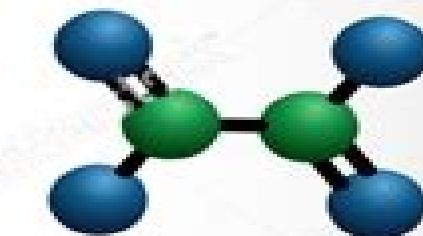
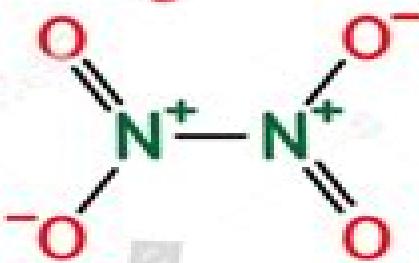
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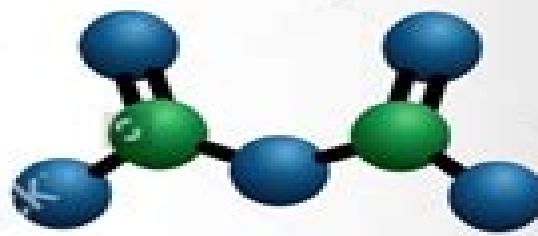
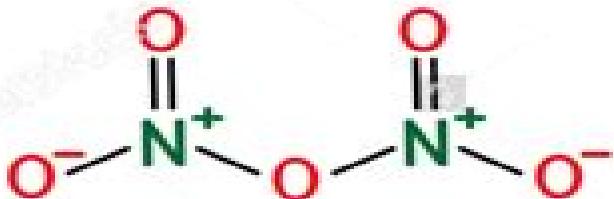
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Oxides Of Nitrogen:

Oxides of Nitrogen C. L. Young, 2013-10-22 Scientists in many fields are interested in the oxides of nitrogen which surround us both in the atmosphere and in the many fluids we encounter daily. This volume comprises experimental solubility data of gaseous nitrous oxide and nitric oxide in all liquids as reported in the scientific literature tables of smoothed mole fraction solubility values for systems which have been studied over a range of temperatures and critical evaluations of the experimental data.

Air Quality Criteria for Oxides of Nitrogen Dennis J. Kotchmar, J. H. Garner, Donald Gardner, Beverly Comfort, 1996-07 Evaluates the latest scientific data on health effects of NOx measured in laboratory animals and exposed human populations and the effects of NOx on agricultural crops, forests and ecosystems as well as the NOx effects on visibility and non-biological materials. Other chapters describe the nature, sources, distribution, measurement and concentrations of NOx in the environment. Covers all pertinent literature through early 1993. Glossary of terms and symbols. Extensive bibliography. Charts, tables and graphs.

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Spectrophotometric Determination of Total Oxides of Nitrogen by Ferrous Sulfate Reaction George Norwitz, FRANKFORD ARSENAL PHILADELPHIA PA QUALITY ASSURANCE

DIRECTORATE, 1965 An accurate and rapid method is proposed for the determination of total oxides of nitrogen in gases from initiators and other explosive devices. Total oxides of nitrogen include nitric oxide, nitrogen dioxide, nitrogen trioxide, nitrogen tetroxide and nitrogen pentoxide. It is frequently customary to determine the total of these oxides because of the

similarity in their toxicity, mode of formation and the ease with which they are converted to each other in equilibrium reactions. In the proposed method the oxides are treated with air and are absorbed into sulfuric acid. 10 to 3 Ferrous sulfate reagent is then added and the pink color is measured. It is not necessary that the air oxidize nitric oxide completely. It is

shown that nitrate and nitrite give the same color with the ferrous sulfate reagent. The method was checked by a technique using a modified Hamilton syringe together with a special gas bulb and laboratory cylinders of nitric oxide and nitrogen

dioxide. Author

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