



# Compounds vs. Mixtures

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Differences Between Solutions and Compounds		
	 <b>Solutions</b>	 <b>Compound</b>
<b>Composition</b>	Made up of substances (elements and compounds) evenly mixed together; the composition can vary in a given mixture.	Made up of atoms bonded together; the combination of atoms is always the same in a given compound.
<b>Changing the composition</b>	The solution is still the same with similar properties. However, the relative amounts of substances might be different.	Changing the composition of a compound has changes into a new compound with new properties.
<b>Properties of parts</b>	The substances keep their own properties when they are mixed.	The properties of the compound are different from the properties of the atoms that make it up.

# Mixtures Compounds And Solutions

**RC Schank**

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*A Treatise on the Theory of Solution, Including the Phenomena of Electrolysis* Sir William Cecil Dampier Dampier, 1902

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