

2025 NFL WEEKLY MATCHUPS SCHEDULE

TEAM	W1	W2	W3	W4	W5	W6	W7	W8	W9	W10	W11	W12	W13	W14	W15	W16	W17	W18
Arizona	@NO	CAR	@SF	SEA	TEN	@IND	GB	BYE	@DAL	@SEA	SF	JAX	@TB	LAR	@HOU	ATL	@CIN	@LAR
Atlanta	TB	@MIN	@CAR	WSH	BYE	BUF	@SF	MIA	@NE	@IND	CAR	@NO	@NYJ	SEA	@TB	@ARI	LAR	NO
Baltimore	@BUF	CLE	DET	@KC	HOU	LAR	BYE	CHI	@MIA	@MIN	@CLE	NYJ	CIN	PIT	@CIN	NE	@GB	@PIT
Buffalo	BAL	@NYJ	MIA	NO	NE	@ATL	BYE	@CAR	KC	@MIA	TB	@HOU	@PIT	CIN	@NE	@CLE	PHI	NYJ
Carolina	@JAX	@ARI	ATL	@NE	MIA	DAL	@NYJ	BUF	@GB	NO	@ATL	@SF	LAR	BYE	@NO	TB	SEA	@TB
Chicago	MIN	@DET	DAL	@LV	BYE	@WSH	NO	@BAL	@CIN	NYG	@MIN	PIT	@PHI	@GB	CLE	GB	@SF	DET
Cincinnati	@CLE	JAX	@MIN	@DEN	DET	@GB	PIT	NYJ	CHI	BYE	@PIT	NE	@BAL	@BUF	BAL	@MIA	ARI	CLE
Cleveland	CIN	@BAL	GB	@DET	MIN	@PIT	MIA	@NE	BYE	@NYJ	BAL	@LV	SF	TEN	@CHI	BUF	PIT	@CIN
Dallas	@PHI	NYG	@CHI	GB	@NYJ	@CAR	WSH	@DEN	ARI	BYE	@LV	PHI	KC	@DET	MIN	LAC	@WSH	@NYG
Denver	TEN	@IND	@LAC	CIN	@PHI	@NYJ	NYG	DAL	@HOU	LV	KC	BYE	@WSH	@LV	GB	JAX	@KC	LAC
Detroit	@GB	CHI	@BAL	CLE	@CIN	@KC	TB	BYE	MIN	@WSH	@PHI	NYG	GB	DAL	@LAR	PIT	@MIN	@CHI
Green Bay	DET	WSH	@CLE	@DAL	BYE	CIN	@ARI	@PIT	CAR	PHI	@NYG	MIN	@DET	CHI	@DEN	@CHI	BAL	@MIN
Houston	@LAR	TB	@JAX	TEN	@BAL	BYE	@SEA	SF	DEN	JAX	@TEN	BUF	@IND	@KC	ARI	LV	@LAC	IND
Indianapolis	MIA	DEN	@TEN	@LAR	LV	ARI	@LAC	TEN	@PIT	ATL	BYE	@KC	HOU	@JAX	@SEA	SF	JAX	@HOU
Jacksonville	CAR	@CIN	HOU	@SF	KC	SEA	LAR	BYE	@LV	@HOU	LAC	@ARI	@TEN	IND	NYJ	@DEN	@IND	TEN
Kansas City	@LAC	PHI	@NYG	BAL	@JAX	DET	LV	WSH	@BUF	BYE	@DEN	IND	@DAL	HOU	LAC	@TEN	DEN	@LV
LA Chargers	KC	@LV	DEN	@NYG	WSH	@MIA	IND	MIN	@TEN	PIT	@JAX	BYE	LV	PHI	@KC	@DAL	HOU	@DEN
LA Rams	HOU	@TEN	@PHI	IND	SF	@BAL	@JAX	BYE	NO	@SF	SEA	TB	@CAR	@ARI	DET	@SEA	@ATL	ARI
Las Vegas	@NE	LAC	@WSH	CHI	@IND	TEN	@KC	BYE	JAX	@DEN	DAL	CLE	@LAC	DEN	@PHI	@HOU	NYG	KC
Miami	@IND	NE	@BUF	NYJ	@CAR	LAC	@CLE	@ATL	BAL	BUF	WSH	BYE	NO	@NYJ	@PIT	CIN	TB	@NE
Minnesota	@CHI	ATL	CIN	@PIT	@CLE	BYE	PHI	@LAC	@DET	BAL	CHI	@GB	@SEA	WSH	@DAL	@NYG	DET	GB
New England	LV	@MIA	PIT	CAR	@BUF	@NO	@TEN	CLE	ATL	@TB	NYJ	@CIN	NYG	BYE	BUF	@BAL	@NYJ	MIA
New Orleans	ARI	SF	@SEA	@BUF	NYG	NE	@CHI	TB	@LAR	@CAR	BYE	ATL	@MIA	@TB	CAR	NYJ	@TEN	@ATL
NY Giants	@WSH	@DAL	KC	LAC	@NO	PHI	@DEN	@PHI	SF	@CHI	GB	@DET	@NE	BYE	WSH	MIN	@LV	DAL
NY Jets	PIT	BUF	@TB	@MIA	DAL	DEN	CAR	@CIN	BYE	CLE	@NE	@BAL	ATL	MIA	@JAX	@NO	NE	@BUF
Philadelphia	DAL	@KC	LAR	@TB	DEN	@NYG	@MIN	NYG	BYE	@GB	DET	@DAL	CHI	@LAC	LV	@WSH	@BUF	WSH
Pittsburgh	@NYJ	SEA	@NE	MIN	BYE	CLE	@CIN	GB	IND	@LAC	CIN	@CHI	BUF	@BAL	MIA	@DET	@CLE	BAL
San Francisco	@SEA	@NO	ARI	JAX	@LAR	@TB	ATL	@HOU	@NYG	LAR	@ARI	CAR	@CLE	BYE	TEN	@IND	CHI	SEA
Seattle	SF	@PIT	NO	@ARI	TB	@JAX	HOU	BYE	@WSH	ARI	@LAR	@TEN	MIN	@ATL	IND	LAR	@CAR	@SF
Tampa Bay	@ATL	@HOU	NYJ	PHI	@SEA	SF	@DET	@NO	BYE	NE	@BUF	@LAR	ARI	NO	ATL	@CAR	@MIA	CAR
Tennessee	@DEN	LAR	IND	@HOU	@ARI	@LV	NE	@IND	LAC	BYE	HOU	SEA	JAX	@CLE	@SF	KC	NO	@JAX
Washington	NYG	@GB	LV	@ATL	@LAC	CHI	@DAL	@KC	SEA	DET	@MIA	BYE	DEN	@MIN	@NYG	PHI	DAL	@PHI

Complete Workbook Nfl Schedule

Mike Meserole

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

temperature lab report temperature measurement 1 studocu - Mar 23 2022

web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees c

experiment 3 lab report chemistry lab 117 experiment 3 pressure - Aug 08 2023

web chemistry lab 117 experiment 3 pressure volume and temperature discovery of the gas laws lab partner amazona tahbou october 8 2014 summary the purpose of this experiment is to be able to understand the relationship between the pressure volume and temperature of a gas understanding this relationship is important as it allows

11 9 the ideal gas law pressure volume temperature and - Apr 04 2023

web the volume of 1.00 mol of any gas at stp standard temperature 273.15 k and pressure 1 atm is measured to be 22.414 l we can substitute 101.325 kpa for pressure 22.414 l for volume and 273.15 k for temperature into the ideal gas equation and solve for r

11 5 charles s law volume and temperature - Feb 19 2022

web may 20 2018 french physicist jacques charles 1746-1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

9 2 relating pressure volume amount and temperature the - Mar 03 2023

web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273.15 k and 1 atm 101.325 kpa 1 at stp one mole of an ideal gas has a volume of about 22.4 l this is referred to as the standard molar volume figure 9.18

lab pressure lab reports lab pressure volume and temperature - Oct 10 2023

web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

8 2 relating pressure volume amount and temperature the - May 05 2023

web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273.15 k and 1 atm or 101.325 kpa to report properties of gases

lab report 5 docx pressure volume and temperature - Aug 28 2022

web pressure volume and temperature discovery of gas laws experiment 5.10.3.19 jared caviglia section 582 ian nicholson introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

pressure volume temperature lab ku school of engineering - Jun 25 2022

web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

lab report pressure volume and temperature discovery - Jan 01 2023

web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

pressure temperature relationship in gases lab studocu - Nov 30 2022

web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

pressure volume temperature pvt laboratory department - Sep 09 2023

web aug 13 2021 pressure volume temperature pvt laboratory analysis of fluid properties such as api gravity viscosity surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and temperature relationships of hydrocarbons are studied at the graduate level

che144 thermodynamics engineering lab report saturation vapor pressure - Jul 27 2022

web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

6 3 relationships among pressure temperature volume and - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts