



2025 NFL WEEKLY MATCHUPS SCHEDULE



TEAM	W1	W2	W3	W4	W5	W6	W7	W8	W9	W10	W11	W12	W13	W14	W15	W16	W17	W18
Arizona	IND	CAR	WSF	SEA	TEN	IND	GB	BYE	GBAL	WSSEA	SF	JAX	ATL	LAR	SHOU	ATL	WSIN	BLAR
Atlanta	TB	WSIN	WCAR	WSH	BYE	SF	WSF	MAA	WSNE	IND	CAR	IND	WSYU	SEA	WSB	WSAR	LAR	NO
Baltimore	WSUF	CLE	DAT	WKC	HOU	LAR	BYE	CHI	WSMA	WSIN	WOLE	WSYU	CHI	PIT	WSIN	NE	WSGB	WPT
Buffalo	BAL	WSYU	MAA	NO	NE	WATL	BYE	WSAR	KC	WSMA	TB	WSHO	WPT	CIN	WSNE	WOLE	PHI	NYJ
Carolina	WSAX	WSAR	ATL	WSNE	WSA	DAL	WSYU	SF	WSB	NO	WATL	WSF	LAR	BYE	WSO	TB	SEA	WSB
Chicago	MIN	WSDET	DAL	WSV	BYE	WSWSH	NO	WSBAL	WSIN	WSYU	WSIN	PIT	WPHI	WSGB	CLE	GB	WSF	DET
Cincinnati	WOLE	JAX	WSIN	WSDN	DET	WSGB	PIT	WSYU	CHI	BYE	WSPE	NE	WSBAL	WSUE	BAL	WSMA	ARI	CLE
Cleveland	CIN	WSBAL	GB	WSCT	MIN	WPT	WSA	WSNE	WSYU	BAL	WSV	SF	TEN	WSCHI	BUF	PIT	WSIN	
Dallas	WPHI	WSG	WSCHI	GB	WSYU	WSAR	WSH	WSDN	ARI	BYE	WSV	Phi	KC	WSDET	MIN	LAC	WSWSH	WSYU
Denver	TEN	IND	WLA	CIN	WPHI	WSO	WSG	DAL	WSHO	LV	KC	BYE	WSWSH	WSV	GB	JAX	WKC	LAC
Detroit	WSB	CHI	WSBAL	CLE	WSIN	WKC	TB	BYE	MIN	WSWSH	WPHI	WSG	GB	DAL	WSAR	PIT	WSIN	WSCHI
Green Bay	DET	WSH	WOLE	WSAL	BYE	CIN	WSAR	WPT	CAR	Phi	WSYC	MIN	WSDET	CHI	WSDN	WSCHI	BAL	WSIN
Houston	WSAR	TB	WSAX	TEN	WSBAL	BYE	WSSEA	SF	DEN	JAX	WSIN	BUF	WSO	WKC	ARI	LV	WLA	IND
Indianapolis	WSA	DEN	WSIN	WSAR	LV	ARI	WLA	TEN	WPT	ATL	BYE	WKC	HOI	WSAX	WSF	JAX	WSHO	
Jacksonville	CAR	WSIN	HOU	WSF	KC	SEA	LAR	BYE	WSV	WSHO	LAC	WSAR	WSIN	IND	NY	WSDN	WSND	TEN
Kansas City	WLA	Phi	WSYC	BAL	WSAX	DET	LV	WSH	WSUF	BYE	WSDN	IND	WSAL	HOI	LAC	WSIN	DEN	WSV
LA Chargers	KC	WSV	DEN	WSYC	WSH	WSMA	IND	MIN	WSIN	PIT	WSAX	BYE	LV	Phi	WKC	WSAL	HOI	WSDN
LA Rams	HOU	WSIN	WPHI	IND	SF	WSBAL	WSAX	BYE	NO	WSF	SEA	TB	WSAR	WSAR	DET	WSSEA	WSATL	ARI
Las Vegas	WSNE	LAC	WSWSH	CHI	WSND	TEN	WKC	BYE	JAX	WSDN	DAL	CLE	WLA	DEN	WPHI	WSHO	WSG	KC
Miami	WSND	NE	WSUF	NYJ	WSAR	LAC	WOLE	WATL	BAL	WSF	WSH	BYE	NO	WSU	WPT	CIN	TB	WSNE
Minnesota	WSCHI	ATL	CIN	WPT	WOLE	BYE	Phi	WLA	WSCT	BAL	CHI	WSGB	WSSEA	WSH	WSAL	WSYC	DET	GB
New England	LV	WSMA	PIT	CAR	WSUF	WSO	WSIN	CLE	ATL	WSB	WSYU	WSIN	WSG	BYE	BUF	WSAL	WSYU	MIN
New Orleans	ARI	SF	WSSEA	WSUF	WSG	NE	WSCHI	TB	WSAR	WSAR	BYE	ATL	WSMA	WSB	CAR	NO	WSIN	WSAL
NY Giants	WSWSH	WSAL	KC	LAC	WSO	Phi	WSIN	WPHI	SF	WSCHI	GB	WSDET	WSNE	WSYU	MIN	WSV	DAL	
NY Jets	PIT	BUF	WSB	WSMA	DAL	DEN	CAR	WSIN	BYE	CLE	WSNE	WSAL	ATL	MIA	WSAX	WSO	NE	WSUF
Philadelphia	DAL	WKC	LAR	WSB	DEN	WSYC	WSIN	WSG	WSYU	BYE	WSB	DET	WSAL	CHI	WLA	LV	WSWSH	WSUF
Pittsburgh	WSYU	SEA	WSNE	MIN	BYE	CLE	WSIN	GB	IND	WSAC	CIN	WSCHI	BUF	WSAL	MIA	WSDET	WOLE	BAL
San Francisco	WSSEA	WSO	ARI	JAX	WSAR	WSB	ATL	WSHO	WSYC	LAR	WSAR	CAR	WOLE	BYE	TEN	WSND	CHI	SEA
Seattle	SF	WPT	NO	WSAR	TB	WSAX	WSH	BYE	WSWSH	ARI	WSAR	WSIN	MIN	WSATL	IND	LAR	WSAR	WSF
Tampa Bay	WATL	WSHO	NYJ	Phi	WSSEA	SF	WSDET	WSO	WSYU	BYE	NE	WSBUF	WSAR	ARI	NO	ATL	WSAR	WSMA
Tennessee	WSDN	LAR	IND	WSHO	WSAR	WSV	NE	WSND	LAC	WSYU	HOI	SEA	JAX	WOLE	WSF	KC	NO	WSAX
Washington	NYC	WSB	LV	WATL	WLA	CHI	WSAL	WKC	SEA	DET	WSMA	BYE	DEN	WSIN	WSYC	Phi	DAL	WPHI

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Mike Miserole

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

experiment 3 lab report chemistry lab 117 experiment 3 pressure - Aug 08 2023

web chemistry lab 117 experiment 3 pressure volume and temperature discovery of the gas laws lab partner amazona tahbou october 8 2014 summary the purpose of this experiment is to be able to understand the relationship between the pressure volume and temperature of a gas understanding this relationship is important as it allows

11 9 the ideal gas law pressure volume temperature and - Apr 04 2023

web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

11 5 charles s law volume and temperature - Feb 19 2022

web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

9 2 relating pressure volume amount and temperature the - Mar 03 2023

web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

lab pressure lab reports lab pressure volume and temperature - Oct 10 2023

web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $PV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

8 2 relating pressure volume amount and temperature the - May 05 2023

web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

lab report 5 docx pressure volume and temperature - Aug 28 2022

web pressure volume and temperature discovery of gas laws experiment 5 10 3 19 jared caviglia section 582 ian nicholson introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

pressure volume temperature lab ku school of engineering - Jun 25 2022

web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

lab report pressure volume and temperature discovery - Jan 01 2023

web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

pressure temperature relationship in gases lab studocu - Nov 30 2022

web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

pressure volume temperature pvt laboratory department - Sep 09 2023

web aug 13 2021 pressure volume temperature pvt laboratory analysis of fluid properties such as api gravity viscosity surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and temperature relationships of hydrocarbons are studied at the graduate level

che144 thermodynamics engineering lab report saturation vapor pressure - Jul 27 2022

web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

6 3 relationships among pressure temperature volume and - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts