



Chemical Interactions

Choose the correct answer:

1. Which is the strongest ionic bond between two atoms, according to the periodic table? (a) Fluorine and Oxygen
(b) Potassium and Chlorine
(c) Chlorine and Oxygen
(d) Potassium and Oxygen

Chlorine and Oxygen

2. Which is the strongest ionic bond with molecules with no charge? (a) Copper and Zinc
(b) Oxygen and Phosphorus
(c) **Gold and Oxygen**
(d) Copper and Zinc

3. Which of the following is an example of a covalent bond? (a) Dissolving a salt
(b) **Combining wood**
(c) Dissolving sugar in water
(d) Dissolving salt

4. Which of the following is an example of a ionic bond? (a) **Combining metals**
(b) Dissolving salt in water
(c) Dissolving sugar in water
(d) Dissolving salt in water

5. Which of the following is commonly known as a ionic bond? (a) Covalent bond
(b) Ionic bond
(c) **None**
(d) Dissolving

6. Which of the following is a sign of a chemical reaction? (a) Dissolving the periodic table
(b) Dissolving the water
(c) **Change in color**
(d) Dissolving the periodic table

7. What do we call the interactions that involve a chemical change? (a) Chemicals
(b) **Chemical**
(c) Chemicals
(d) Phosphorus

8. Which of the following is NOT a type of chemical bond? (a) Metalloid bond
(b) Dissolving water
(c) **Electrostatic bonds**
(d) Dissolving salt

9. Which is the primary physical reaction that is used to test for acids and bases? (a) Electrolysis
(b) Dissolving
(c) **Titration**
(d) Dissolving

10. Which type of reaction involves the breakdown of a substance into smaller substances? (a) **Combining**
(b) Dissolving the periodic table
(c) Dissolving the water
(d) Dissolving the periodic table

Multiple Chemical Interactions

KJ Lindholm-Leary

Multiple Chemical Interactions:

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