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# Dangerous Vintage

Laura Pender



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country, the sweetest  
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## Other Laura Harlequin Intrigue No 367

**C Cleary**

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid h<sub>2</sub>co<sub>3</sub> and its conjugate base the bicarbonate ion hco<sub>3</sub><sup>-</sup>

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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