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# Self Help 2026 Guide

**Donna D. Ignatavicius, Tami K. Little**



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web you may have to consult the solubility rules write the net ionic equation for the reaction of  $\text{BaCl}_2 \text{ (aq)}$  and  $\text{Na}_2\text{SO}_4 \text{ (aq)}$  you may have to consult the solubility rules write

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web  $\text{CuClO}_4$  2 is soluble all perchlorates are soluble while  $\text{PbSO}_4$  is insoluble sulfates are generally soluble except for  $\text{Pb}^{2+}$   $\text{Ba}^{2+}$   $\text{Hg}_2^{2+}$  and  $\text{Sr}^{2+}$  the net ionic equation is

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