



THE  
**MATCHCOVER  
COLLECTOR'S  
PRICE GUIDE**

2nd Edition

by  
**Bill Retskin**

with contributions by  
**John Williams**

THE COMPREHENSIVE REFERENCE BOOK  
AND PRICE GUIDE TO MATCHCOVERS

# Matchcover Collectors Price Guide

**Kyle Husfloen**

## **Matchcover Collectors Price Guide:**

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categories is better than ever Includes listings for collector clubs specialty periodicals dealers collectors experts buyers appraisers parts suppliers reproduction sources Internet resources repair restoration conservation specialists auction services manufacturers distributors of contemporary collectibles and more    **Cigar Aficionado** ,1997-08    **Antique Trader's Cashing in Your Collectibles** Miriam Plans,1999 A resource for identifying a variety of objects determining the basis of interest and their value and deciding whether to sell them and how to choose the proper outlets for disposing of those items    Trash Or Treas GD of Buyers Tony Hyman,2000    *Trash Or Treasure* Tony Hyman,1994    Kovels' Yellow Pages Ralph M. Kovel,Terry H. Kovel,1999 Need an eye for your teddy bear Searching for another soup spoon to complete Grandma s silver Looking for someone to repair your porcelain doll Wondering if restoring your carousel horse will lower its value Kovels Yellow Pages is the ultimate directory for collectors It helps you find sources for repair restoration and parts and it lists matching services appraisers and auction houses all you need to help you maintain the value of your antiques and add to your enjoyment of collecting America s antiques experts Ralph and Terry Kovel also give you insider information on how to navigate the complicated world of buying selling and collecting antiques They have distilled a lifetime s worth of information to give you the last word on Where to start when you ve inherited a house full of stuff How to know if you need an appraiser and where to find one When giving away an antique is more profitable than selling it If repairing or restoring an object will lower or enhance its value How to bid in out of town auctions Kovels Yellow Pages gives you firsthand information from more than 3 000 suppliers clubs auctions services and industry sources nationwide You ll find names addresses telephone and fax numbers plus e mail and Internet addresses There s also an extensive bibliography of price books and reference books most helpful to collectors With this guide you have at your fingertips the most complete and up to date collector s reference available anywhere    **The Official Rinker Price Guide to Collectibles** Harry L. Rinker,2000-10-10 This fourth edition of the Harry L Rinker Official Price Guide to Collectibles has more coverage of collectibles than any other book on the market Here you ll find furniture decorative accessories and giftware along with the traditional character and personality items ceramic glass and toys Each category includes a brief history collecting tips reference books periodicals collector clubs and vital information on reproductions It s a complete document of the 20th century American lifestyle    *Signatures of the Stars* Kevin Martin,1998 An alphabetical listing of Hollywood stars collectible television casts and rock and roll stars with a sample autograph very brief biography range of prices and scarcity index    *Encyclopedia of Associations* ,2004    **Warman's Antiques and Their Prices** Edwin G. Warman,1992

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web  $m \times c \times \Delta t$  equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction  $\Delta h_{soln}$  is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as  $NaOH(s) \rightarrow Na(aq) + OH(aq)$   $\Delta h_{soln} = 44.2 \text{ kJ mol rxn}$  write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta h_{soln}$  or the amount of energy released or absorbed in kj mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

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web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer

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