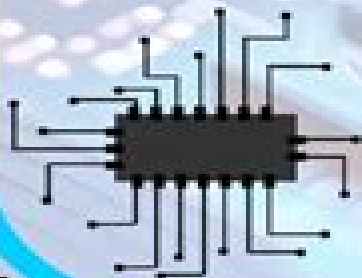


Metals

- Metals are materials known for their strength, conductivity, and malleability
- They are structured in a lattice of cations surrounded by a 'sea' of delocalized electrons, which allows them to conduct electricity and heat efficiently
- Metals can be alloyed to enhance properties like strength and corrosion resistance
- Commonly used in construction, electronics, and manufacturing, metals are integral to modern society

Practical example: Electronics

- Metals are crucial in electronics due to their electrical conductivity
- Components like wires and circuits are primarily made of metals such as copper and aluminum, which allow electricity to flow efficiently
- The metal's ability to conduct electricity is essential for the functioning of any electronic device, from smartphones to computers
- Their durability and resistance to corrosion also make metals indispensable in long-lasting electronic applications



Modern Metals Science In Action

Anthony S. Fauci



Modern Metals Science In Action:

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