

MODERN METHODS OF CHEMICAL ANALYSIS

Introduction, general methods of quantitative analysis, advantages and disadvantages of the methods, principles of classical methods and instrumental methods, spectroscopic methods of analysis – properties of EMR, EM spectrum, interaction of EMR with matter and its phenomenon, various types of spectra, laws of spectrophotometry – Beer Lambert's law, its derivation, deviations of Beer Lambert's law, Applications of Beer Lambert's law in qualitative and quantitative analysis, numerical problems.

(4)

Introduction:

A qualitative method gives information about the identity of atomic or molecular or ionic species or the functional groups present in the sample. A quantitative method provides numerical information of the constituents i.e., the relative amounts of one or more of the constituents of the sample. The quantitative methods are often classified as either classical or instrumental/modern. The classical methods are also called as wet-chemical methods.

Classical and modern/instrumental methods of analysis:

In the early years of chemistry, most analyses (which were being analyzed) were carried out by separating the components of interest by precipitation, extraction or distillation. For qualitative analysis, the separated components were then treated with reagents that yielded products that could be identified by their colors, boiling or melting points, solubilities in a series of solvents, odors, optical activities or their refractive indices. These are the classical quantitative methods, which are the volumetry and gravimetry.

The phenomena other than those used for classical methods were exploited for solving the analytical problems in early twentieth century. The measurement of physical properties of analyte such as conductivity, electrode potential, light absorption or emission, mass-to-charge ratio, fluorescence etc. were used for quantitative analysis. The techniques like distillation, extraction and precipitation etc. were replaced by high efficient chromatography and electrophoretic techniques for separation of components of complex mixtures. These newer methods for separation and determination of chemical species are collectively known as instrumental methods of analysis.

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