

# MOLE CONCEPT

Scientific notation  
 $N \times 10^n \rightarrow$  Integer  
 $N < 10$

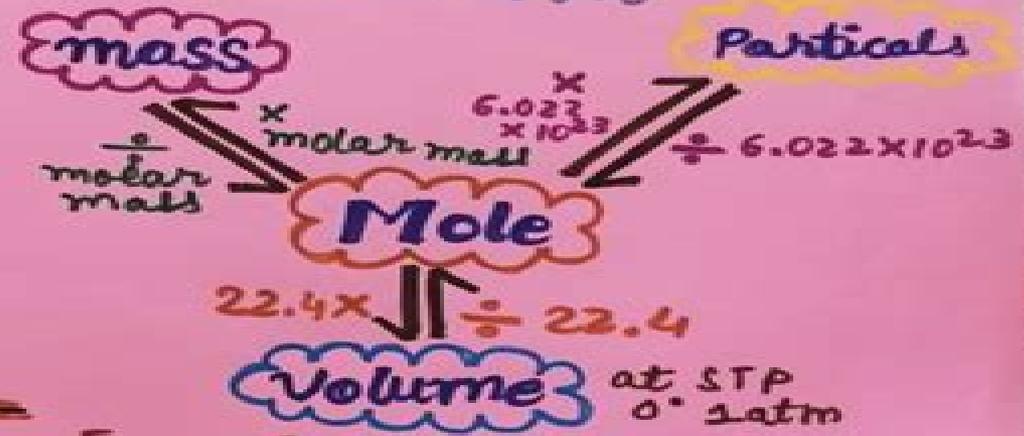
Density  $\rightarrow \frac{\text{mass}}{\text{volume}}$   
 Temperature  $\rightarrow$   
 $^{\circ}\text{F} = \frac{9}{5}(^{\circ}\text{C}) + 32$   
 $\text{K} = ^{\circ}\text{C} + 273.15$

Limiting Reagent  
 The reactant which present in less quantity. gets consumed after sometime. by this we given mole can find LR.  
 moles

Isotopes - Z same  
 A diff  
 Isobars - Z diff  
 A same  
 Isoelectronic - Same num e<sup>s</sup>

molar mass - mass of one mole substance in grams.  
 one mole: -  $6.022 \times 10^{23}$  particles.  
 percentage composition: -  $\frac{\text{mass}_c}{\text{molar mass}_c} \times 100$   
 Empirical formula = molecular formula

## Conversions



\* Mass per cent  
 $= \frac{\text{mass of solute} \times 100}{\text{mass of solution}}$

\* Mole fraction  
 $= \frac{\text{No. of moles of A}}{\text{No. of moles of solution}}$

\* Molarity  
 $(M) = \frac{\text{No. of mole of solute}}{\text{Volume of solution (lit)}}$

\* Molality  
 $(m) = \frac{\text{no. of moles of solute}}{\text{mass of solvent (in kg)}}$

Empirical formula - gives simplest whole number ratio of various elements in one molecule of compound.  
 formula of  $\text{C}_6\text{H}_{12}\text{O}_6$   
 Empirical for.  $\text{CH}_2\text{O}$

Molecular formula - gives actual ratio of atom of various element in one molecule of com.  
 $\text{MF} = n \times \text{EF}$   
 n is a common factor



# Mole Notes

**Michael Brown**



## **Mole Notes:**

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American Notes and Queries William Shepard Walsh, Henry Collins Walsh, William H. Garrison, Samuel R. Harris, 1889  
**A Revision of the American Moles** Frederick William True, 1896     **Notes on Dental Anatomy and Physiology and Dental Histology** Thomas William Widdowson, 1928     The Weekly Notes Frederick Pollock, 1905     *NBS Technical Note* ,1973-08     **The poetical works of John Milton, ed. with notes by J. Bradshaw** John Milton, 1885     Technical Note ,1921     Shakespeare's Hamlet, with notes, examination papers, and plan of preparation, ed. by J.M.D. Meiklejohn William Shakespeare, 1880     Notes and Queries: a Medium of Inter-communication for Literary Men, Artists, Antiquaries, Genealogists, Etc ,1873     *Concise Notes in Physical Chemistry* Robert Sydney Lowrie, 1965     **Weekly Notes of Cases Argued and Determined in the Supreme Court of Pennsylvania, the County Courts of Philadelphia, and the United States District and Circuit Courts for the Eastern District of Pennsylvania** ,1878     *Stationery World and Printing and Allied Trades* ,1918     **Wild Mammals of North America** George A. Feldhamer, Bruce C. Thompson, Joseph A. Chapman, 2003-11-19 Table of contents     Proceedings of the United States National Museum United States National Museum, 1897     **Bulletin** United States. Department of Agriculture. Office of Publications, 1896

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