

Interpreting the Periodic Table

Examine the hypothetical periodic table below and answer the given questions.

A 10x10 grid with various shaded regions and labels. The grid is composed of 100 squares. Shaded regions include a 2x3 rectangle in the top-left, a 3x2 rectangle in the top-right, a 3x3 square in the center-right, a 2x2 square in the center-right, a 2x2 square in the bottom-right, and a 2x2 square in the bottom-right. Labels are placed in specific squares: 'A' in the bottom-left, 'B' in the bottom-right, 'C' in the center-left, 'D' in the center-right, 'E' in the top-center, 'F' in the middle-center, 'G' in the top-right, 'H' in the middle-right, 'I' in the bottom-center, 'K' in the top-left, 'L' in the top-right, 'P' in the bottom-right, and 'Z' in the bottom-right.

- ① Which pair of elements has the same number of valence electrons?

② Which pair of metals are in the same period? _____

③ Which element has the smallest atomic number? _____

④ Which elements are metals? _____

⑤ Which elements are nonmetals? _____

⑥ If the atomic number of the element P is 41, what is the atomic number of the element P?

Mastering The Periodic Table Exercises On The Elements

Susan Wise Bauer, Jessie Wise

Mastering The Periodic Table Exercises On The Elements:

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