

MOLE CONCEPT

Scientific notation
 $N \times 10^n \rightarrow$ Integer
 $N < 10$

Density $\rightarrow \frac{\text{mass}}{\text{volume}}$
 Temperature \rightarrow
 $^{\circ}\text{F} = \frac{9}{5}(^{\circ}\text{C}) + 32$
 $\text{K} = ^{\circ}\text{C} + 273.15$

Limiting Reagent

The reactant which present in less quantity. gets consumed after sometime. by this we given mole can find LR.

Isotopes - Z same
 A diff
 Isobars - Z diff
 A same
 Isoelectronic - Same num e's

molar mass - mass of one mole substance in grams.

one mole: - 6.022×10^{23} particles.

percentage composition: - $\frac{\text{mass}_c}{\text{molar mass}_c} \times 100$

Empirical formula = molecular formula

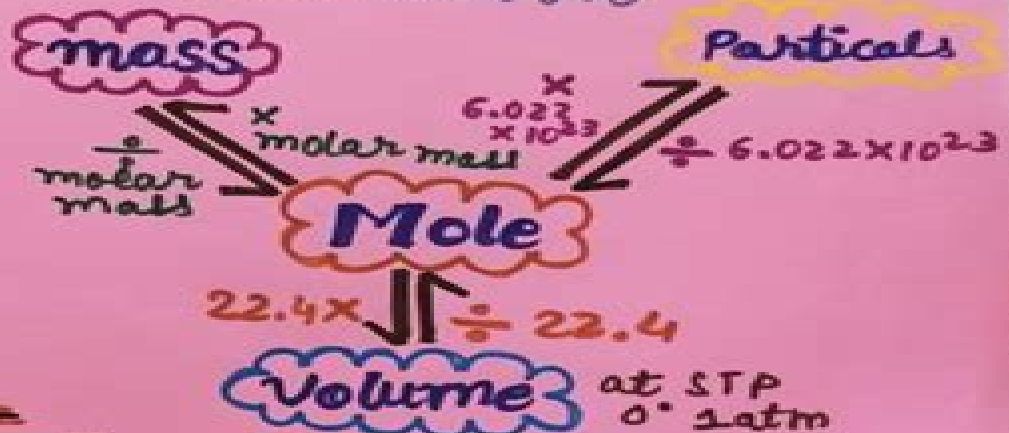
* Mass per cent
 $= \frac{\text{mass of solute} \times 100}{\text{mass of solution}}$

* Mole fraction
 $= \frac{\text{No. of moles of A}}{\text{No. of moles of solution}}$

* Molarity
 $(M) = \frac{\text{No. of mole of solute}}{\text{Volume of solution (lit)}}$

* Molality
 $(m) = \frac{\text{no. of moles of solute}}{\text{mass of solvent (in kg)}}$

Conversions



Empirical formula: - gives simplest whole number ratio of various elements in one molecule of compound.
 formula of $\text{C}_2\text{H}_{12}\text{O}_6$
 Empirical for. CH_2O

Molecular formula: - gives actual ratio of atom of various element in one molecule of compound.
 $\text{MF} = n \times \text{EF}$
 n is a common factor



Mole Notes

Robert Sydney Lowrie



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In a world inundated with monitors and the cacophony of instantaneous transmission, the profound energy and psychological resonance of verbal artistry often disappear into obscurity, eclipsed by the constant onslaught of sound and distractions. However, nestled within the musical pages of **Mole Notes**, a charming perform of fictional splendor that impulses with raw feelings, lies an unique trip waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that exciting opus manuals readers on a mental odyssey, softly exposing the latent potential and profound affect stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is key subjects, dissect its captivating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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